

Name _____
Date _____

Wavelength, Frequency, Quantum and Average Atomic Mass Wkst

1. What is the wavelength of a light with the frequency of 1.81×10^{-14} Hz?
2. What is the frequency of a light with the wavelength of 680nm?
3. What is the energy of the light wave in problem #2?
4. What is the energy of light with the wavelength 1230nm?
5. Calculate the energy of a quantum of radiant energy, whose frequency is 3.82×10^{-14} Hz. Show work!
6. According to the formula $c =$ _____, as frequency gets larger (higher) the wavelength gets _____
7. What is the energy of light with the wavelength 732 nm(nanometers)?
8. A certain light has a wavelength of 923 nm. What is its frequency?
9. What is the energy of light with the wavelength of 943nm?
10. What is the frequency of a light with the wavelength of 860nm?
11. What is the energy of the light wave in # 10?