

Common Polyatomic Ions

1- charge		2- charge		3- charge	
Formula	Name	Formula	Name	Formula	Name
H_2PO_4^-	Dihydrogen phosphate	HPO_4^{2-}	Hydrogen phosphate	PO_4^{3-}	Phosphate
$\text{C}_2\text{H}_3\text{O}_2^-$	Acetate	$\text{C}_2\text{O}_4^{2-}$	Oxalate	PO_3^{3-}	Phosphite
HSO_3^-	Hydrogen sulfite	SO_3^{2-}	Sulfite		
HSO_4^-	Hydrogen sulfate	SO_4^{2-}	Sulfate		
HCO_3^-	Hydrogen carbonate	CO_3^{2-}	Carbonate		
NO_2^-	Nitrite	CrO_4^{2-}	Chromate		
NO_3^-	Nitrate	$\text{Cr}_2\text{O}_7^{2-}$	Dichromate	1+ charge	
CN^-	Cyanide	SiO_3^{2-}	Silicate	Formula	Name
OH^-	Hydroxide			NH_4^+	Ammonium
MnO_4^-	Permanganate				
ClO^-	Hypochlorite				
ClO_2^-	Chlorite				
ClO_3^-	Chlorate				
ClO_4^-	Perchlorate				